Turnover Rate & Residence Time (Mini-Activity)

**Purpose**
- To understand turnover rate and residence time, in the context of the global carbon cycle.

**Overview**
Students discuss as a class the concepts of turnover rate and residence time using a simplified example. Students use the *Global Carbon Cycle Diagram* to calculate turnover rate and residence time for each pool.

**Questions**

**Background**

**Pool** (also stock or reservoir): A pool is the storehouse of material in a portion of the environment. Examples of 'pools' scientists might consider include: carbon in leaves, trees or entire ecosystems; water in a river, lake or all of the world’s oceans; calcium in rocks, seashells or your own body. Scientists use the concept of a pool as a way of simplifying what would otherwise be very difficult to study.

**Turnover rate**: The fraction of material that leaves a pool in a specified time interval. Turnover rate is the mathematical inverse of residence time.

**Residence time**: The average length of time that material spends in a given pool. Residence time depends on the rate of outflow and on the size of the pool. Residence time is the mathematical inverse of turnover rate.

**What To Do and How To Do It**

**EXPLORE**

**Grouping:** Class/Small Groups  **Time:** 40 minutes

- Begin by addressing the definitions of and differences between turnover rate and residence time.
- Go over Example 1 (below) as a class to provide a concrete experience.
- Have students try Example 2 on their own, and then go over it as a class.
- Provide each student with the Turnover Rate & Residence Time worksheet and ask them to work in pairs to complete the table and answer the follow-up questions.
  - You may need/want to go over which fluxes are the inputs/outputs to each carbon pool before they begin their work.
Assessment

• Did students calculate turnover rate and residence time correctly?
• Look for evidence of students’ critical and creative thinking in their responses to the follow-up questions.

Adaptations

• Some students may need more than just equations to understand these concepts. Set up a classroom demonstration where you manipulate objects (such as colored beads in a beaker) so students can see the movement of material into and out of the pool.
  • Start with 20 blue beads in a beaker (pool). Tell students that you will remove 5 blue beads per cycle and you will add 5 red beads per cycle.
  • After 4 cycles students will observe that 100% of the pool as been turned over.
  • \( \frac{5 \text{ beads/cycle}}{20 \text{ beads}} = 0.25 \text{, } 20 \text{ beads/(5 beads/cycle)} = 4 \text{ cycles} \)

References

  File: Carbon Budget 2009 Presentation [2010, November 15].

Example 1 – A High School:

Consider a high school that has 800 students (pool) and these students are evenly divided among the 4 grade levels (Freshman, Sophomore, Junior, Senior). If we assume that all seniors graduate every spring, 200 students are leaving the school (output flux). Let us also assume 200 new freshman enter each fall (input flux). (Thus our system is in equilibrium.)

If turnover rate is the fraction of students that leave the school each year (output flux/pool size), then our equation is:

\[
\frac{200 \text{ students/year}}{800 \text{ students}} = 0.25 \text{ per year}
\]

This means that 25% of the student body is graduating and leaving the school each year.

If residence time is the number of years that a student spends at the school before they graduate (pool size/output flux), then our equation is:

\[
\frac{800 \text{ students}}{200 \text{ students/year}} = 4 \text{ years}
\]

This means that every student spends 4 years in the school before they graduate.
Example 2 – A Small Lake:

The lake is the pool. It contains 1000 liters of water. Water is what we are interested in. The total flow of water into the lake is 40 liters per year. This is the input. The total flow of water out of the lake is 40 liters per year. This is the output.

Calculate the turnover rate and residence time of the lake:

**Turnover rate** is the fraction of water that leaves the lake each year.

\[
\frac{40 \text{L/year}}{1000 \text{L}} = 0.04 \text{ per year}
\]

This calculation tells us that 4% of the water leaves the lake each year, which means that the turnover rate of water in the lake is 4% per year.

**Residence time** is the average length of time water spends in the lake. Residence time is the ratio of the pool size, to the rate of outflow.

\[
\frac{1000 \text{L}}{40 \text{L/year}} = 25 \text{ years}
\]

Residence time is also the inverse of turnover rate:

\[
\frac{1}{0.04 \text{/year}} = 25 \text{ years}
\]

Remember if there are multiple outputs you will need to sum them to get the total amount of material leaving the pool over a 1-year time span.
Turnover Rate & Residence Time

Using the *Global Carbon Cycle Diagram*, calculate turnover and residence time for all carbon pools in the global carbon cycle.

**Notes** – Use outputs to calculate turnover rate and residence time. If a pool has multiple outputs, sum them before making calculations.

<table>
<thead>
<tr>
<th>Pool</th>
<th>Turnover Rate</th>
<th>Residence Time</th>
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</thead>
<tbody>
<tr>
<td>Atmosphere 750Pg C</td>
<td>Photosynthesis: 120 + Ocean Uptake: 92 = 212Pg C/year</td>
<td>750/212 = 3.5 years</td>
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<tr>
<td></td>
<td>212/750 = 0.28 = 28%/year</td>
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<tr>
<td>Earth’s Crust 100,000,000Pg C</td>
<td>Volcanos: 0.1Pg C/year</td>
<td>1000000000/0.1 = 1 billion years</td>
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<td></td>
<td>0.1/1000000000 = 1^-9 = 0.0000001%/year</td>
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<tr>
<td>Oceans 38,000Pg C</td>
<td>Ocean Loss: 90 + Burial to Sediment: 0.1 = 90.1Pg C/year</td>
<td>38000/90.1 = 422 years</td>
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<td>90.1/38000 = 0.002 = 2%/year</td>
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<tr>
<td>Plants 560Pg C</td>
<td>Respiration: 59 + Litterfall: 59 + Land Use Change 2.0= 120Pg C/year</td>
<td>560/120 = 4.7 years</td>
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<td></td>
<td>120/560 = 0.21 = 21%/year</td>
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<tr>
<td>Soils 1,500Pg C</td>
<td>Soil Respiration: 58Pg C/year</td>
<td>1500/58 = 26 years</td>
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<td>58/1500 =0.039 = 3.9%/year</td>
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<tr>
<td>Fossil Fuels 7,500Pg C</td>
<td>Burning Fossil Fuels: 6.3Pg C/year</td>
<td>7500/6.3 = 1190 years</td>
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<td>6.3/7500 = 0.00084 = 0.084%/year</td>
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Follow-up questions: (You are looking for evidence of students’ critical and creative thinking.)
1) Do you think the residence time of carbon in the fossil fuel pool is realistic? Why or why not?

While for this particular time period the residence time may be realistic this scenario assumes that both the rate of fossil fuel burning and the size of the fossil fuel pool are not changing over time. Already, since 1995, as cited by Schlesinger (1997), the rate of fossil fuel burning has increased from 6PgC/year to 7.7PgC/year (Global Carbon Project, 2010) and is continuing to rise. In addition to an increase in fossil fuel burning flux, it is also important to realize that the fossil fuel pool is a finite resource (because there are no new inflows). Although new fossil fuels can form, the rate is significantly slower than the rate at which they are being used. For this reason, fossil fuels are considered to be a limited resource. (In contrast, the plant pool is constantly dying and re-growing at a similar rate.) If you were to try and calculate a new residence time of carbon in the fossil fuel pool, you would need to predict the rate of burning and know the new fossil fuel pool size.

2) Why do you think it is important to understand turnover rate and residence time in the context of the global carbon cycle?

An understanding of turnover rates and residence times is essential for understanding how the materials in different parts of our environment are changing. With respect to carbon, this is very important because of the effect carbon in the atmosphere has on the Earth’s climate. Because the components of a system are all interconnected, a change in any carbon pool can lead to a change in how much carbon is in the atmosphere. In light of the relationship between the carbon cycle and climate change, scientists may ask questions such as: Is there a way to increase residence time in the soil or terrestrial vegetation? Will there be a feedback between global temperatures and the ability of the ocean to store carbon? Will warmer temperatures increase decomposition, thus accelerating the rate at which carbon is transferred from soils to the atmosphere (i.e. reducing the residence time of the soil carbon pool)? Industry specialists may want to calculate: How long will the Earth’s fossil fuel reserves last? Will there be enough to continue business as usual?
Turnover Rate & Residence Time

Using the Global Carbon Cycle Diagram, calculate turnover and residence time for all carbon pools in the global carbon cycle.

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Follow-up Questions: Use critical thinking, there is more than one correct answer.

1) Do you think the residence time of carbon in the fossil fuel pool is realistic? Why or why not?

2) Why do you think it is important to understand turnover rate and residence time in the context of the global carbon cycle?